

IN SILICO INVESTIGATION OF XANTHONE DERIVATIVE POTENCY IN INHIBITING CARBONIC ANHYDRASE II (CA II) USING MOLECULAR DOCKING AND MOLECULAR DYNAMICS (MD) SIMULATION

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ABSTRACT

Objective: Hypertension is the leading contributor to all-cause death and disability worldwide. One of the most well-known first-line antihypertensive drugs is chlorthalidone which treats hypertension through carbonic anhydrase (CA) II inhibition. However, due to the high number of cases of hypertension, a more potent medication is still needed. Xanthone is a potential candidate for the compound group for its potency in inhibiting CA II. Therefore, this research aims to evaluate around 500 xanthonenes' potency as a better oral antihypertensive drug than chlorthalidone.

Methods: 507 xanthonenes were analyzed for their potency using *in silico* method. Xanthone's structures were retrieved from the PubChem website or built using Avogadro software, while the CA II receptor was retrieved from The RCSB website. Then molecular docking, ADME evaluation, and toxicity test were evaluated from selected ligands. Finally, a molecular dynamics simulation was conducted to evaluate the stability of the potential ligand as the inhibitor of CA II protein.

Results: This research found that globulixanthone c is considered to be a better CA II inhibitor compared to chlorthalidone. It is due to its lower binding affinity compared to chlorthalidone and its stable binding to CA II's important inhibition sites with low fluctuation. It also has the potential to be consumed orally because it fulfills all of Lipinski's rule of five standards and its toxicity is on the moderate level.

Conclusion: Globulixanthone c, a type of prenylated xanthonenes group, showed the best potential activity as the inhibitor of CA II protein to treat hypertension among other xanthonenes.

Keywords: Carbonic anhydrase (CA) II, Hypertension, Xanthone, Molecular docking, Molecular dynamics

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INTRODUCTION

Systemic arterial hypertension (or hypertension) is a condition characterized by persistently high blood pressure (BP) in the systemic arteries. BP is often stated as the ratio of the systolic BP (that is, the pressure that blood exerts on the arterial walls when the heart contracts) and the diastolic BP (the pressure when the heart relaxes) [1]. The Seventh Report of the Joint National Committee on Prevention, Detection, Evaluation, and Treatment of High Blood Pressure (JNC) 7 categorizes the disease as a condition when one's systolic/diastolic pressure exceeds 140/90 mmHg [2].

Hypertension becomes the most common risk factor for cardiovascular disease (CVD), chronic kidney disease (CKD), and cognitive impairment. It is the single leading contributor to all-cause death and disability worldwide [1]. According to WHO (2021), 1.13 of 7.8 billion of the world's population are diagnosed with hypertension, and the disease itself is estimated to cause 7.5 million deaths (which is 12.8% of total deaths) worldwide [3, 4]. The high level of hypertension cases and deaths caused by it make hypertension treatment become crucial to lower the world's morbidity level and increase global life expectancy. Some researchers conducted research on targeting certain proteins involved in hypertension using natural products and a standard drug to elaborate the activity and potential pathway inhibition. Demir (2019) revealed the potential inhibitor of some antihypertensive drugs to PON1 protein that links with another disease [5]. In addition, Demir (2020) also showed the potential of quinones as an antihypertensive agent [6].

One of the most common first-line antihypertensive medications is thiazide diuretics with chlorthalidone as the most commonly used

drug in the group [7]. In lowering blood pressure, chlorthalidone's important mechanism is carbonic anhydrase (CA) inhibition [8]. Carbonic anhydrase (EC 4.2.1.1) is a group of metalloenzymes that catalyzes the hydration of CO₂ and H₂O into bicarbonate and hydrogen ions [9]. This enzyme plays a role in various physiological processes in various organisms, including humans, therefore, the abnormal level or activity of this enzyme can trigger diseases [9, 10]. Agree with that, inhibition of this enzyme becomes crucial to treat some diseases and to examine the new potential drug for clinical applications [11]. In the human body, there are 13 catalytically active CA isozymes that spread in various concentrations and locations [8]. These are classified according to crucial properties such as inhibitor sensitivity, catalytic activity, and subcellular location [12]. Amongst these isozymes, isozymes I, II, III, IV, V, IX, XII, and XIV have relevance in cardiovascular regulation. From them, CA II is counted as a very potent drug target to lower blood pressure because of its high activity (10⁵-6/sec) and its various locations—red cells, kidney, lung, heart, brain, vascular smooth muscle, and endothelium. However, some research revealed the side effects of thiazide diuretics, especially from chlorthalidone treatment, such as hypokalemia, hyperglycemia, myocardial infarction, hospitalization for heart failure, ischemic or hemorrhagic stroke, and a composite cardiovascular disease [13, 14]. Therefore, the necessity of looking for a new hypertension treatment is still required.

Xanthonenes as a group of secondary metabolites are normally found in a restricted assembly of higher plants (mostly family Clusiaceae and Gentianaceae), fungi, and lichens. This compound group has a symmetrical parent compound—9H-xanthen-9-one and was classified into six groups—simple xanthonenes, glycosylated xanthonenes, prenylated xanthonenes, xanthonolignoids, bis-xanthonenes, and miscellaneous

xanthenes [15]. Xanthenes and their derivatives have broad biological activity, especially in the medical field [16]. Their antioxidant activity is good for antimicrobial agents [17]. Furthermore, Abuzaid *et al.* [18] revealed xanthone's benefit for preventing obesity and metabolic syndrome. Some xanthone derivatives were found to be good inhibitors of CA II. One of them is mangiferin, a type of glycosylated xanthenes, which was found by Saleem *et al.* to be a good inhibitor of CA II due to its good inhibition concentration of the enzyme using both molecular docking and *in vitro* methods [19]. Furthermore, Davis *et al.* [20] showed that xanthenes extraction from microfungus of the genus *Xylaria* had potential inhibitors for carbonic anhydrase enzymes and had better inhibition activity than phenol extract. This finding made xanthone deserve to be researched further, corresponding to its potential to be a better CA II inhibitor.

At present, around 500 xanthenes have been reported in the previous research. However, there is no research found that compares the ability among around 500 xanthenes to act as the best CA II protein inhibitor. Therefore, this study investigated 507 xanthone compounds' molecular interaction and dynamics against CA II by molecular docking and molecular dynamics (MD) simulations approach using chlorthalidone as the standard inhibitor.

MATERIALS AND METHODS

Ligand retrieval and preparation

The 507 xanthone compounds and standard ligand in the 3D structure were gathered. 332 of the xanthone compounds and chlorthalidone (PubChem ID: 2732) 3D structures were retrieved in 3D. sdf format from PubChem (<https://pubchem.ncbi.nlm.nih.gov/>) while 175 of the rest were made using Avogadro software. The ligands retrieved from PubChem were given CHARMM force field and MMFF94 partial charge and converted to a. pdb file using Discovery Studio 2020 (BIOVIA). On the other hand, the ligands made using Avogadro software were given MMFF94 force field and saved in. sdf format. Later, the ligands made manually were opened in Discovery Studio 2020 (BIOVIA), given CHARMM force field and MMFF94 partial charge, and saved in. pdb format. These ligands were then brought further into the next screening.

Protein retrieval and preparation

The protein used as the target for this study is CA II. The crystal structure of CA II was retrieved from Protein Data Bank (PDB) (<https://www.rcsb.org/>) with the PDB ID: 1I91. The retrieved crystal structure was then prepared using Discovery Studio 2020 (BIOVIA) by removing water molecules, ligand molecules, and ions.

Table 1: Protein target, their binding sites, their grid settings for docking

Protein target	Center	Dimension (Å)
CA II (PDB ID: 1I91)	X: -7.107	X: 25
	Y: -0.669	Y: 25
	Z: 12.624	Z: 25

Ligand filtering using lipinski's rule of five

Ligand filtering was done based on the ligands' fulfillment of Lipinski's rule of five (Ro5). The Absorption, Distribution, Metabolism, and Excretion (ADME) properties of the 507 ligands were firstly retrieved from the SwissADME (<http://www.swissadme.ch/>) web tool by inputting the SMILES string of the ligands. The SMILES strings of the 332 ligands from PubChem were also obtained from the site. Whereas the SMILES strings of the manually made 175 ligands were obtained from ChemBD (<http://chemdb.ics.uci.edu/cgi-bin/BabelWeb.py>) web tool by inputting the. sdf format of the ligands. After obtaining the SMILES strings of the ligands, the strings were then inputted into SwissADME. Only the ligands that pass all of the Ro5 would be brought to the next screening.

Molecular docking validation

Before molecular docking was done, the docking method was previously validated. If the RMSD is below 3 Å, then the method is

acceptable [21]. The validation was done by re-docking the crystal native ligand (6-[n-(3-hydroxy-phenyl)-3-(morpholin-4-ylmethyl)-2h-thieno[3,2-E]-1,2-thiazine-1,1-dioxide]-sulfonamide) to the prepared protein CA II using the PyRx 0.9.8 software [22]. The native ligand's post-docking conformation as compared to its crystal structure conformation RMSD using Pymol software.

Molecular docking

After the docking method validation, the prepared protein and ligands were loaded to PyRx 0.9.8 and then prepared to be docked in the built-in AutoDock VINA in the PyRx program [22]. The docking done in this research was specific-site docking with grid selection parameters as shown in table 1 and the rest of the setting was left as default. The binding affinity result table and the best model of each protein-ligand interaction with the five most negative binding affinities were saved to be visualized. Docking visualization was done using Discovery Studio 2020 (BIOVIA) to see the 2D and 3D interactions of each protein-ligand's best models. Docking analysis and visualization were done in Microsoft Windows PC with Intel® Core™ i5-8265U CPU @1.60GHz 1.80 GHz, 8 GB RAM with NVIDIA GeForce MX230 ver. 462.31.

Ligand toxicity test

Ligands with the five most negative binding affinities of each protein were then evaluated, corresponding to their toxicity score. The toxicity test used in this research is based on oral rat LD₅₀ score using ADMETlab (https://admet.scbdd.com/calcpred/calcf_single_mol/#) web tool by inputting the ligands' SMILES strings into the site.

Conserved amino acid analysis

To analyze the conserved amino acid region, the ConSurf web server was used (<https://consurf.tau.ac.il/>). PDB ID 1I91 was uploaded to the server and Multiple Sequence Alignment (MSA) was used for this analysis. The conserved region was displayed using a gradation color scale from 1 as the variable region to 9 as the conserved region.

Molecular dynamics simulation

The best ligand from the previous filtering then simulated molecular dynamically using GROMACS and followed the MD simulation [23]. The ligand-protein post-dock conformation was firstly prepared as input. The preparation started by separating the ligand-protein file into protein and ligand files and saving them in the. pdb format. Then, the topologies and post-processed. gro files of both files were made. To create the protein topology and post-processed. gro file, the addition of CHARMM36 force field and TIP3P water model and ignoring of H atoms were done to the protein. pdb file. On the other hand, to create the ligand topology and post-processed. gro file, the addition of H atoms and CGenFF force field were done to the ligand. pdb file. After the system's energy was minimized, the system was equilibrated. The equilibration was done in two phases-NVT and NPT phases. After setup, the system was then equilibrated in the NVT, then NPT phase, for 500 ps for each phase. In the NVT phase, the system was equilibrated to reach 309.5 degrees Celsius. After that, in the NPT phase, the system was equilibrated to reach a stable density.

RESULTS AND DISCUSSION

Ligand filtering using lipinski's rule of five (Ro5)

After 507 of the ligands, 3D structures and SMILES were collected and filtered. In this filtering, it is found that only 412 ligands fulfill all Lipinski's rule of five-100% (146 of 146 ligands) from simple xanthenes, 5% (3 of 60 ligands) from glycosylated xanthenes, 91.81% (258 of 281 ligands) from prenylated xanthenes, 100% (2 of 2 ligands) from xanthonolignoids, 0% (0 of 11 ligands) from bis-xanthenes, and 42.9% (3 of 7 ligands) from miscellaneous xanthenes (table S1).

Lipinski's rule of five (Ro5) is a methodology to set the drugability properties for drug formulation [24]. Compounds that fulfill Lipinski's rule of five are predicted to have favorable oral bioavailability and drug-like characteristics [25]. Among four rules in the Ro5, molecular weight (MW) and H-bond donor/acceptor

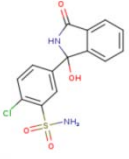
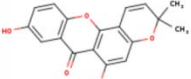
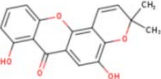
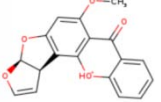
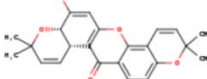
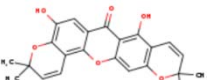
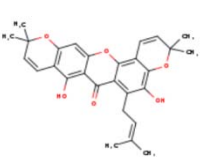
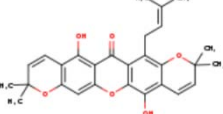
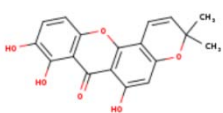
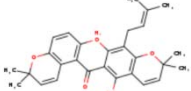
were the main challenges in filtering the xanthone compounds (table S1). Molecular weight is important in drug development as higher MW will tend to have higher lipophilicity characteristics [26]. Furthermore, a proper balance between lipophilicity and hydrophilicity is also crucial in drug design. Excessing hydrogen donor/acceptor can disrupt the hydrophilicity balance by decreasing the affinity of the hydrophobic region [27]. According to table 2, simple xanthenes and prenylated xanthenes groups were the

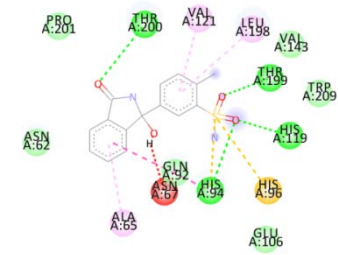
xanthone group that most fulfil Ro5 requirements. Simple xanthenes only consist of simple substituents such as hydroxy, methoxy, or methyl group with around three benzene rings as the main structure, similar to prenylated xanthenes. Conversely, Bis-xanthenes are the xanthone group that consists of more than one cluster benzene ring with more hydroxyl and carbonyl residue [15]. It brings to the more complex structure with excess MW and hydrogen donor/acceptor requirements from Ro5.

Table 2: Ligands with 0 lipinski's rule of five violation

Xanthone group	Total ligands	Total ligands with 0 lipinski's rule of five violation
Simple xanthenes	146	146
Glycosylated xanthenes	60	3
Prenylated xanthenes	281	258
Xanthonolignoids	2	2
Bis-xanthenes	11	0
Miscellaneous Xanthenes	7	3
Total	507	412

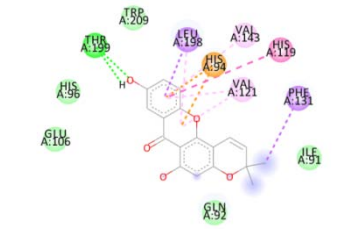
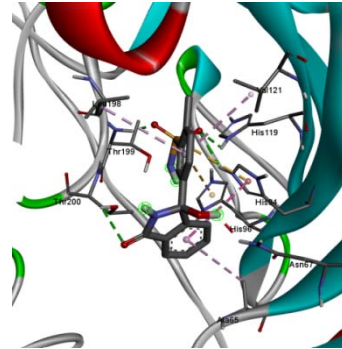
Table 3: Ligands with 5 lowest docking binding affinity with CA II

Ligand	2D structure	Binding affinity (kcal/mol)
Chlorthalidone (standard ligand)		-8.2
Nigrolineaxanthone F		-9
Nigrolineaxanthone H		-9
7-Deoxysterigmatocystin		-9.1
Brasilixanthone A		-9.1
Nigrolineaxanthone I		-9.2
Mangostenone A		-9.2
Garcimangosone A		-9.5
Globulixanthone C		-9.5
Calophinone		-9.8



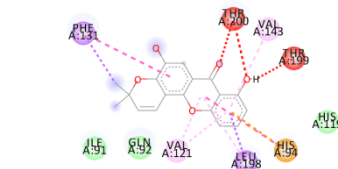
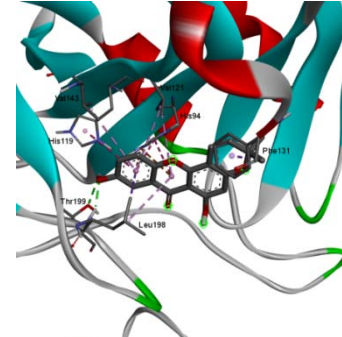
Interactions
 van der Waals
 Conventional Hydrogen Bond
 Unfavorable Donor-Donor
 Pi-Sulfur
 Pi-Pi T-shaped
 Pi-Alkyl

(a) chlorthalidone



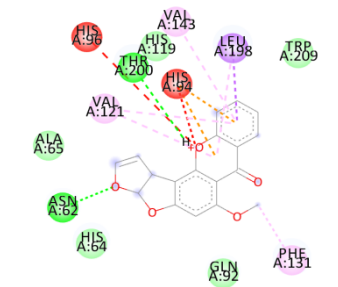
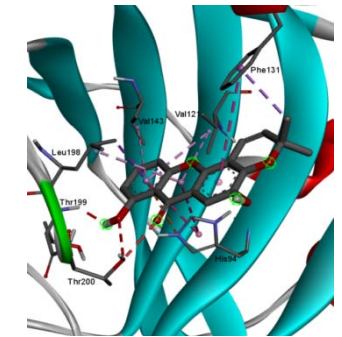
Interactions
 van der Waals
 Conventional Hydrogen Bond
 Pi-Cation
 Pi-Sigma
 Pi-Pi Stacked
 Pi-Pi T-shaped
 Pi-Alkyl

(b) nigrolineaxanthone f



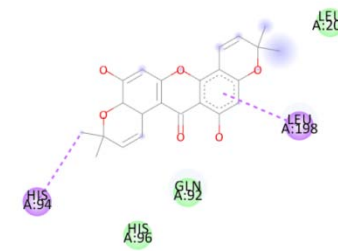
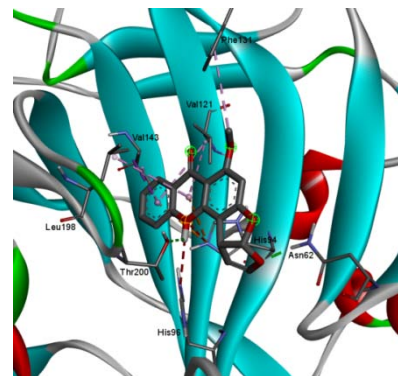
Interactions
 van der Waals
 Unfavorable Donor-Donor
 Unfavorable Acceptor-Acceptor
 Pi-Cation
 Pi-Sigma
 Pi-Pi Stacked
 Pi-Pi T-shaped
 Pi-Alkyl

(c) nigrolineaxanthone h



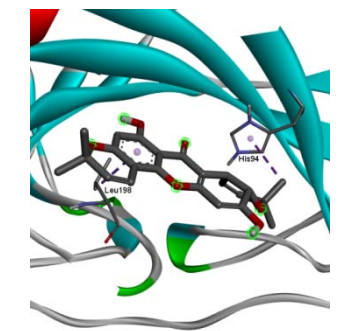
Interactions
 van der Waals
 Conventional Hydrogen Bond
 Unfavorable Positive-Positive
 Pi-Cation
 Pi-Sigma
 Pi-Alkyl

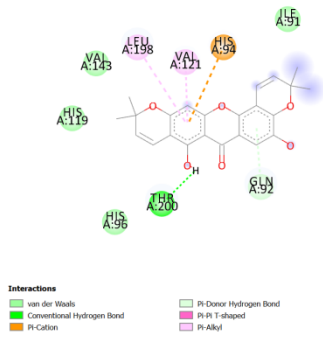
(d) 7-deoxysterigmatocystin



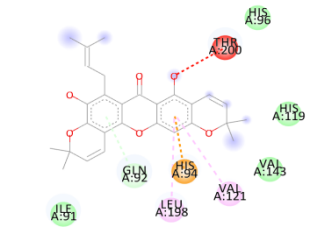
Interactions
 van der Waals
 Pi-Sigma

(e) brasilixanthone a

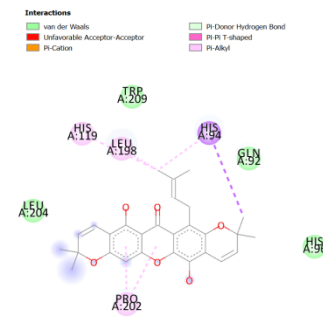




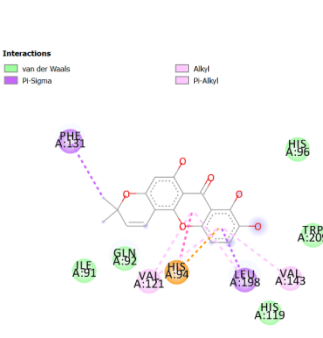
(f) nigrolineaxanthone i



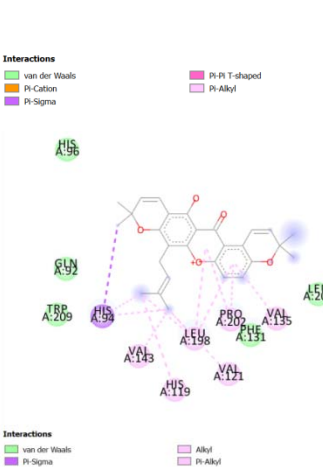
(g) mangostenone a



(h) garcimangosone a



(i) globulixanthone c



(j) calophinone

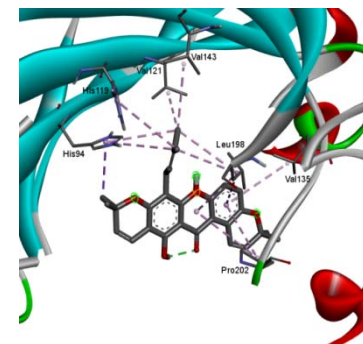
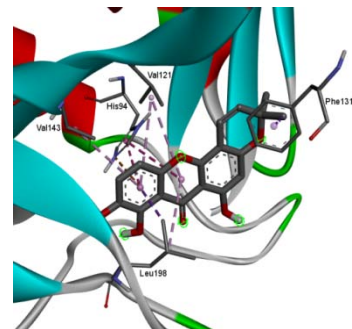
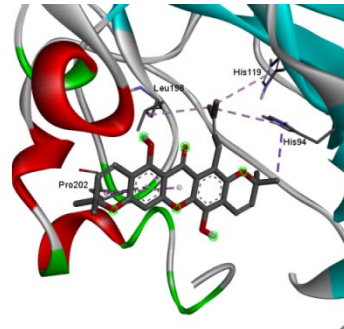
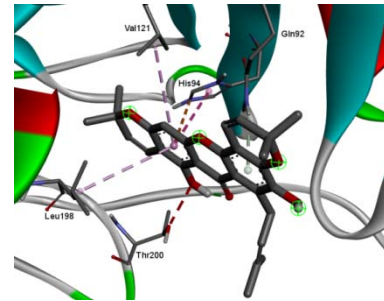
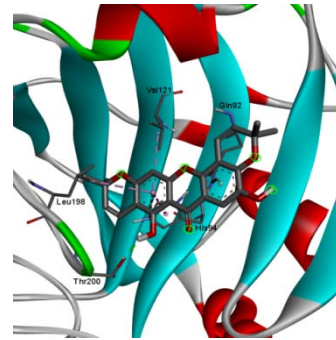


Fig. 3: Post-dock 2D and 3D interactions of (a) chlorthalidone, (b) nigrolineaxanthone f, (c) nigrolineaxanthone h, (d) 7-deoxysterigmatocystin, (e) brasilixanthone a, (f) nigrolineaxanthone i, (g) mangostenone a, (h) garcimangosone a, (i) globulixanthone c, (j) calophinone with CA II

Molecular docking

Molecular docking validation was conducted to validate the docking method. The result of RMSD was 2,6 Å and it was acceptable to continue the docking process (fig. S1) [21]. The docking of the 412 ligands that fulfill all of the Ro5 and the standard ligand-chlorthalidone-to CA II was done. The binding site for specific docking and the grid settings were set up as shown in table 1.

After the 412 ligands were docked to CA II, it was found that 85 of the ligands have a more negative binding affinity than chlorthalidone, which indicates that the ligands then have a better binding to CA II than chlorthalidone. Chlorthalidone's binding affinity itself is -8.2 kcal/mol, while the binding affinity range of the ligands with lower binding affinity than chlorthalidone is -8.3 kcal/mol to -9.8 kcal/mol (table S2)

The 85 ligands were then filtered further and the ligands with the 5 lowest binding affinities were chosen to be evaluated more and brought to the next screening. Nine (9) ligands were found to have the 5 lowest docking binding affinity with CA II shown in table 3–nigrolineaxanthone f (-9 kcal/mol), nigrolineaxanthone h (-9 kcal/mol), 7-deoxysterigmatocystin (-9.1 kcal/mol), brasilixanthone a (-9.1 kcal/mol), nigrolineaxanthone i (-9.2 kcal/mol), mangostenone a (-9.2 kcal/mol), garcimangosone a (-9.5 kcal/mol), globulixanthone c (-9.5 kcal/mol), and calophinone (-9.8 kcal/mol).

As we can observe in fig. 3, chlorthalidone makes hydrogen bonding with His94, His119, Thr199, and Thr200, pi interactions with Ala65, Val121 and Leu198 (pi-alkyl), His94 and His96 (pi-sulfur), and His94 (pi-pi T-shaped), and unfavorable interaction with Asn67 (donor-donor). Hydrogen bonding is a strong non-covalent bond that occurs when hydrogen that is covalently bonded with a very electronegative atom (N, O, or F) is attracted by the electrons of another atom nearby it. Whereas pi interaction is relatively weaker than the hydrogen bond that occurs between molecules with a pi system from conjugated molecules like benzene.

The same hydrogen bonding sites as chlorthalidone were found to be interacting with some of the test ligands, namely nigrolineaxanthone f (with Thr199 two times), 7-deoxysterigmatocystin (with Thr200), and nigrolineaxanthone i (with Thr200). Besides, the same pi-alkyl interaction sites were also found to interact in pi-alkyl interaction, alkyl or pi-sigma with the test ligands–nigrolineaxanthone f (with Val121 two times and Leu198 (pi-alkyl) and with Leu198 (pi-sigma)), nigrolineaxanthone h (with Val121 two times and Leu198 (pi-alkyl) and with Leu198 (pi-sigma)), 7-deoxysterigmatocystin (with Val121 two times and Leu198 (pi-alkyl) and with Leu198 (pi-sigma)), brasilixanthone a (with Leu198 (pi-sigma)) nigrolineaxanthone i (with Val121 and Leu198 (pi-alkyl)), mangostenone a (with Val121 and Leu198 (pi-alkyl) and with Leu198 (alkyl)), globulixanthone c (with Val121 two times and Leu198 (pi-alkyl) and with Leu198 (pi-sigma)), and calophinone (with Leu198 two times (pi-alkyl) and with Val121 and Leu198 (alkyl)).

Other pi interaction sites in chlorthalidone–pi-sulfur interaction sites (His94 and His96)–were not found to interact in pi-sulfur interaction with the test ligands. However, His94 (which is also the

site of pi-pi T-shaped interaction with chlorthalidone) makes other forms of pi interactions with the test ligands, that is, nigrolineaxanthone f (in pi-pi T-shaped and pi-cation interactions (two times both), nigrolineaxanthone h (in pi-pi T-shaped (two times) and pi-cation interactions), 7-deoxysterigmatocystin (in pi-cation interactions (two times)), brasilixanthone a (in pi-sigma interaction), nigrolineaxanthone i (in pi-pi T-shaped and pi-cation interactions), nigrolineaxanthone i (in pi-pi T-shaped and pi-cation interactions), mangostenone a (in pi-pi T-shaped and pi-cation interactions), garcimangosone a (in pi-sigma interaction), globulixanthone c (in pi-pi T-shaped (two times) and pi-cation interactions) and calophinone (in pi-sigma interaction)

According to the previous research [28-30] Asn62, Ala65, His94, His96, Val121, Phe131, Leu141, Val143, Leu198, Thr199, Thr200, Val207, Trp209 were the important residues to stabilize CAII inhibitor. Chlorthalidone and 7-deoxysterigmatocystin were built with 9 important residues out of 13 in total. They interacted with various chemical bondings such as hydrogen, hydrophobic, and van der waals interactions. Furthermore, other ligands are only built with 3 to 7 important residues with almost similar chemical bonding interactions. Simone *et al.* [31] stated that strong CAII inhibition was correlated with liposolubility characteristics. Agree with that, all 9 potential ligands showed high liposolubility based on the MLOGP value on the Ro5 [table S1]. Interestingly, the important residues in the CAII protein show a hydrophobicity region (fig. S2) and it will increase the chance to interact with the nonpolar molecular surfaces of the potential ligands. Freitas *et al.* [32] also stated that hydrophobic interaction was crucial to developing high-efficiency ligands. Hydrophobic interaction is shown as alkyl, pi-alkyl, pi-pi stacked, pi-pi t shaped, and pi-sigma interaction. Based on table 3, all potential ligands showed various hydrophobic interactions.

Ligand oral rat LD₅₀ toxicity test

After the selected ligands were docked to CA II, the oral rat LD₅₀ toxicity test was done on the selected ligands. This toxicity test states that a compound with a certain amount of its dose is categorized into some level of toxicity if it causes the death of one-half of a group of test animals (rats) that consumed the compound orally. The oral rat LD₅₀ score of a compound is often expressed as mg/kg, with mg stating the amount of the compound and the kg stating the weight of the consumer.

According to Hodge and Sterner's scale of toxicity, a compound is categorized into some levels of toxicity, that is, extremely toxic if its oral rat LD₅₀ score is <1 mg/kg, highly toxic if its score is 1–5 mg/kg, moderately toxic (50–500 mg/kg), slightly toxic (500–5,000 mg/kg), practically non-toxic (5,000–15,000 mg/kg), and relatively harmless (>15,000 mg/kg).

The selected ligands were found to have toxicity levels around moderately toxic and slightly toxic. As shown in table 5, among the selected ligands, only nigrolineaxanthone h is found to be labeled with a slightly toxic toxicity level. The rest of the ligands are all categorized as moderately toxic toxicity. For further analysis, globuloxanthone C was selected according to the binding affinity score, important residues interaction, and toxicity result (table 6).

Table 5: Selected xanthone's oral rat LD₅₀ toxicity score and their oral rat LD₅₀ toxicity level according to hodge and sterner

Ligand	Oral rat LD ₅₀ toxicity score (mg/kg)	Hodge and sterner's oral rat LD50 toxicity level
Chlorthalidone	2,264.35	slightly toxic
Nigrolineaxanthone F	462.15	moderately toxic
Nigrolineaxanthone H	509.1	slightly toxic
7-Deoxysterigmatocystin	246.81	moderately toxic
Brasilixanthone A	424.58	moderately toxic
Nigrolineaxanthone I	486.1	moderately toxic
Mangostenone A	251.32	moderately toxic
Garcimangosone A	247.87	moderately toxic
Globulixanthone C	465.18	moderately toxic
Calophinone	225.88	moderately toxic

Table 6: Selected ligand parameters

Ligands	Parameters			
	Binding affinity (kcal/mol)	Important residues interaction	Hydrophobic interaction	Toxicity result (mg/kg)
Chlorthalidone (control)	-8.3	9	3	2264
Nigrolineaxanthone F	-9	7	4	462
Nigrolineaxanthone H	-9	5	4	509
7-Deoxysterigmatocystin	-9.1	9	4	246
Brasilixanthone A	-9.1	3	2	424
Nigrolineaxanthone I	-9.2	6	2	486
Mangostenone A	-9.2	6	2	251
Garcimangosone A	-9.5	3	4	247
Globulixanthone C	-9.5	7	4	464
Calophinone	-9.8	7	7	225

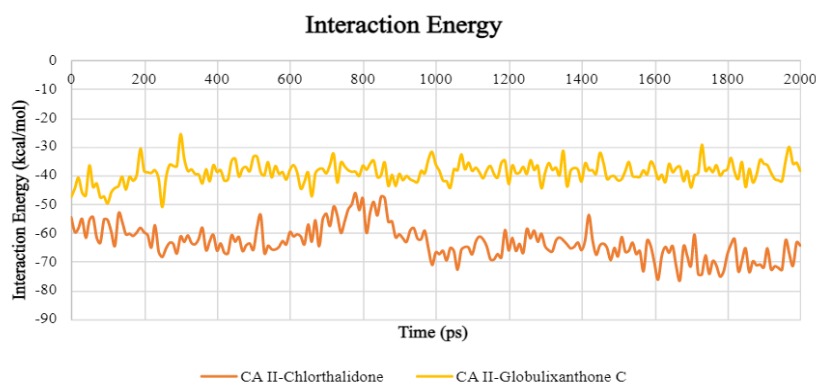


Fig. 7: CA II-chlorthalidone and CA II-globulixanthone c interaction energy fluctuation in 2 ns simulation

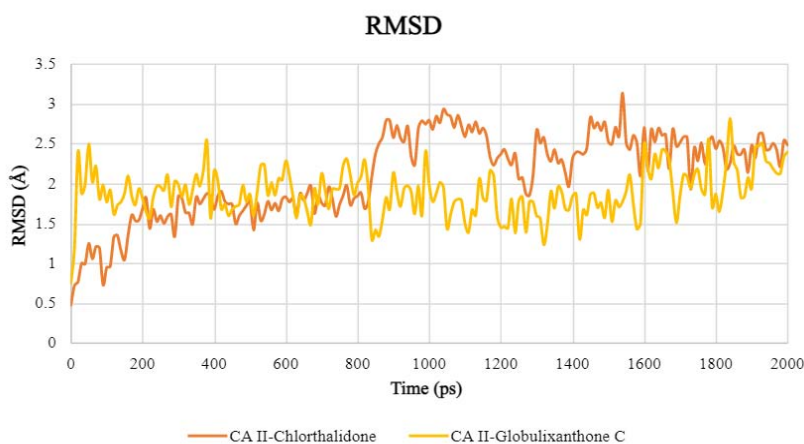


Fig. 8: CA II-chlorthalidone and CA II-globulixanthone c RMSD fluctuation in 2 ns simulation

Molecular dynamics simulation

Globulixanthone c, as the selected ligand, was then simulated in MD simulation with CA II. After the systems were equilibrated, they both run in the simulation for 2 ns. The interaction energy, RMSD, important interactions, and RMSF were evaluated. The interaction energy used in this analysis is the total of Lennard-Jones and Coulombic energy of the protein and the ligand binding. Therefore, this interaction energy is not enough to depict the overall binding of the ligand to the protein. However, this interaction energy can be used to see how the ligand poses and binding to the protein changes over time.

CA II-chlorthalidone's average interaction energy is -63.39 kcal/mol while for CA II-globulixanthone c is -39.04 kcal/mol (fig. 7). As depicted in fig. 7, the CA II-globulixanthone c complex's binding is more stable than the standard chlorthalidone with CA II complex.

The binding stability can also be seen from the Root Mean Square Deviation (RMSD) fluctuation of the complexes. RMSD is a measurement of how the ligand's position in binding to its target changes along simulation compared to its starting structure [33, 34]. CA II-globulixanthone c's average RMSD is shown to be lower -1.88 Å than CA II-chlorthalidone's average RMSD -2.11 Å. CA II-globulixanthone c's RMSD fluctuation is also more stable compared to CA II-chlorthalidone's fluctuation. This indicates that in the interaction energy and RMSD, globulixanthone c could be a better inhibitor for CA II compared to the standard drug chlorthalidone.

We also evaluate the CA II-chlorthalidone and CA II-globulixanthone c complexes for Root Mean Square Fluctuation (RMSF) result. RMSF is a measurement of how much each of the protein residues fluctuates within the simulation. The larger the RMSF score means the higher the flexibility and instability of the residue. While the

smaller it is, then the lower the flexibility and instability of the residue [35]. From the RMSF result, both complexes showed an

almost similar pattern suggesting that globulixanthone c has good potential for CAII inhibitor.

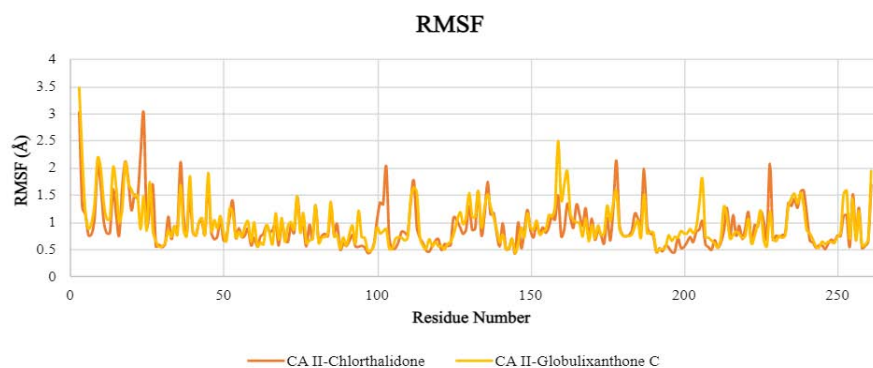


Fig. 9: CA II-chlorthalidone and CA II-globulixanthone c RMSF in 2 ns simulation

Table 8: RMSF values of important CA II residues when simulated with chlorthalidone and globulixanthone c

Residues	RMSF (Å)	
	CA II-chlorthalidone	CA II-globulixanthone c
Asn62	0.7	0.6
Ala65	0.8	0.8
His94	0.6	1.2
His96	0.5	0.7
Val121	0.5	0.6
Phe131	0.9	1.1
Leu141	1.0	0.8
Val143	0.5	0.5
Leu198	0.7	0.7
Thr199	0.5	0.8
Thr200	0.5	0.8
Val207	0.6	0.7
Trp209	0.5	0.7

For deeper analysis, RMSF evaluation was also conducted on the important residues of CA II-Asn62, Ala65, His94, His96, Val121, Phe131, Leu 141, Val 143, Leu198, Thr199 and Thr200, Val207, Trp209 when simulated to globulixanthone c compared to chlorthalidone. The RMSF of the residues are shown in table 8 and the result showed that the important residues are in a similar range from both complexes and it was still considered stable for the RMSFs as the value was around 1 Å. Furthermore, the important residues were also analyzed for their conservation level. Amitai *et al.* [36] and Buyong *et al.* [37] stated that the important residues play a crucial role in drug development, especially for ligand binding. Fig. S3 showed that these residues reveal a high conservation scale compared to other organisms.

CONCLUSION

Globulixanthone c, a type of prenylated xanthone, is considered to be a potential CA II inhibitor candidate due to its lower binding affinity (-9.5 kcal/mol) than the standard drug chlorthalidone (-8.2 kcal/mol) from molecular docking result and its stable binding to CA II's important inhibition sites from MD simulation. It also has the potential to be consumed orally because it fulfills all of Lipinski's rule of five and its toxicity is at a moderate level.

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AUTHORS CONTRIBUTIONS

All authors have contributed equally.

CONFLICT OF INTERESTS

Among the authors have no conflict of interest

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